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Note: Not all of the reactions will occur. For those that do not, write no reaction.

1. A piece of copper is dropped into a container of water.  
No reaction

2. Liquid bromine is added to a container of sodium iodide crystals.  $\text{Br (l)} +$

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$2\text{NaI}(s) + 2\text{NaBr}(s) +$   
 $12(s)$  3. An  
aluminum strip is  
immersed in a  
solution of silver  
nitrate.

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tin(II) carbonate  
sodium hydrogen  
carbonate NaHCO  
1. 2. 3. 4. 5. 6. 8. 9.

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10. 11.

vanadium(V) oxide

H<sub>2</sub>O dihydrogen

monoxide (NH) CO

ammonium oxalate

4 224 polonium(VI)

thiocyanate

Po(SCN)<sub>6</sub>

tetraphosphorus

decaoxide P<sub>4</sub>O<sub>10</sub>

12. 13. 15. 16. 17.

18. 19.

manganese(VII)

oxide copper(II)

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series, predict and  
balance the  
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reactions.

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