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of Aqueous Solutions and Precipitation Reactions The Ultimate Chemical Equations Handbook

Note: Not all of the reactions will occur. For those that do not, write no reaction.

1. A piece of copper is dropped into a container of water. No reaction
2. Liquid bromine is added to a container of sodium iodide crystals. $\text{Br (l)} + 2\text{NaI(s)} \rightarrow 2\text{NaBr(s)} + \text{I}_2\text{(s)}$
3. An aluminum strip is immersed in a solution of silver nitrate.

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The Ultimate Chemical Equations Handbook SnCO tin(II) carbonate sodium hydrogen carbonate NaHCO 1. 2. 3. 4. 5. 6. 8. 9. 10. 11. vanadium(V) oxide H₂O dihydrogen monoxide

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(NH) CO ammonium oxalate 4 224 polonium(VI) thiocyanate
Po(SCN)₆ tetraphosphorus decaoxide P₄O₁₀ 12. 13. 15.
16. 17. 18. 19. manganese(VII) oxide copper(II) dihydrogen
phosphate

East Boston High School

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