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An up-to-date introduction to the field, treating in depth the electronic structures of atoms, molecules, solids and surfaces, together with brief descriptions of inverse photoemission, spin-polarized photoemission and photoelectron diffraction. Experimental aspects are considered throughout and the results carefully interpreted by theory. A wealth of measured data is presented in tabullar for easy use by experimentalists.

This text presents a balanced presentation of the macroscopic view of empirical kinetics and the microscopic molecular viewpoint of chemical dynamics. This second edition includes the latest information, as well as new topics such as heterogeneous reactions in atmospheric chemistry, reactant product imaging, and molecular dynamics of H + H2.

Laboratory experiences as a part of most U.S. high school science curricula have been taken for granted for decades, but they have rarely been carefully examined. What do they contribute to science learning? What can they contribute to science learning? What is the current status of labs in our nation's high schools as a context for learning science? This book looks at a range of questions about how laboratory experiences fit into U.S. high schools: What is effective laboratory teaching? What does research tell us about learning in high school science labs? How should student learning in laboratory experiences be assessed? Do all student have access to laboratory experiences? What changes need to be made to improve laboratory experiences for high school students? How can school organization contribute to effective laboratory teaching? *With* increased attention to the U.S. education system and student outcomes, no part of the high school curriculum should escape scrutiny. This timely book investigates factors that influence a high school laboratory experience, looking closely at what currently takes place and what the goals of those experiences are and should be. Science educators, school administrators, policy makers, and parents will all benefit from a better understanding of the need for laboratory experiences to be an integral part of the science curriculum and how that can be accomplished.

EXPERIMENTS IN GENERAL CHEMISTRY, Sixth Edition, has been designed to stimulate curiosity and insight, and to clearly connect lecture and laboratory concepts and techniques. To accomplish this goal, an extensive effort has been made to develop experiments that maximize a discovery-oriented approach and minimize personal hazards and ecological impact. Like earlier editions, the use of chromates, barium, lead, mercury, and nickel salts has been avoided. The absence of these hazardous substances should minimize disposal problems and costs. This lab manual focuses not only on what happens during chemical reactions, but also helps students understand why chemical reactions occur. The sequence of experiments has been refined to follow topics covered in most general chemistry textbooks. In addition, Murov has included a correlation chart that links the experiments in the manual to the corresponding chapter topics in several Cengage Learning general chemistry titles. Each experiment--framed by pre-and post-laboratory exercises and concluding thought-provoking questions--helps to enhance students' conceptual understanding. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

The demonstrations capture interest, teach, inform, fascinate, amaze, and perhaps, most importantly, involve students in chemistry. Nowhere else will you find books that answer, "How come it happens? . . . Is it safe? . . . What do I do with all the stuff when the demo is over?" Shkhashiri and his collaborators offer 282 chemical demonstrations arranged in 11 chapters. Each demonstration includes seven sections: a brief summary, a materials list, a step-by-step account of procedures to be used, an explanation of the hazards involved, information on how to store or dispose of the chemicals used, a discussion of the phenomena displayed and principles illustrated by the demonstration, and a list of references.

The collection of contributions in this volume presents the most up-to-date findings in catalytic hydrogenation. The individual chapters have been written by 36 top specialists each of whom has achieved a remarkable depth of coverage when dealing with his particular topic. In addition to detailed treatment of the most recent problems connected with catalytic hydrogenations, the book also contains a number of previously unpublished results obtained either by the authors themselves or within the organizations to which they are affiliated. Because of its topical and original character, the book provides a wealth of information which will be invaluable not only to researchers and technicians dealing with hydrogenation, but also to all those concerned with homogeneous and heterogeneous catalysis, organic technology, petrochemistry and chemical engineering.

Oxidizing and Reducing Agents S. D. Burke University of Wisconsin at Madison, USA R. L. Danheiser Massachusetts Institute of Technology, Cambridge, USA Recognising the critical need for bringing a handy reference work that deals with the most popular reagents in synthesis to the laboratory of practising organic chemists, the Editors of the acclaimed Encyclopedia of Reagents for Organic Synthesis (EROS) have selected the most important and useful reagents employed in contemporary organic synthesis. Handbook of Reagents for Organic Synthesis: Oxidizing and Reducing Agents, provides the synthetic chemist with a convenient compendium of information concentrating on the most important and frequently employed reagents for the oxidation and reduction of organic compounds, extracted and updated from EROS. The inclusion of a bibliography of reviews and monographs, a compilation of Organic Syntheses procedures with tested experimental details and references to oxidizing and reducing agents will ensure that this handbook is both comprehensive and convenient.

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